

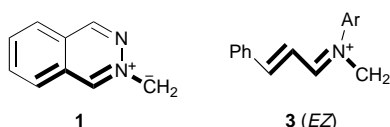
Stereoisomeric Styryl-substituted Pyrrolidines, 3,7-Diazabicyclo[3.3.0]octanes and 2-Styrylpyrroles from Cinnamaldehyde Iminium-*N*-methanide 1,3-Dipoles†

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The reaction of cinnamaldehyde *N*-aryliminium methanide 1,3-dipoles with dimethyl maleate, dimethyl fumarate and some *N*-aryl-maleimides gives equal mixtures of stereoisomeric substituted pyrrolidines and a route to 2-styrylpyrroles with dimethyl acetylenedicarboxylate.

Azomethine methanide (ylide) 1,3-dipoles have proved to be important synthons for the pyrrole system.^{1–3} Routes to these dipoles include thermal ring-opening of aziridines,^{1–4} reactions of α -amino esters with carbonyl compounds,^{5–6} reactions of amines with aldehydes⁷ and desilylation of trimethylsilylmethyliminium salts.^{8,9} We have recently examined¹⁰ the stereochemistry of the cycloadducts of the phthalazinium-2-methanide 1,3-dipole **1**. This contains a more rigid structural component than the analogous cinnamaldehyde iminium-methanide dipole **3** and we wished to compare the behaviour of **1** with this more flexible structure **3**.



In the generation of 1,3-dipoles from iminium salts the possible *EZ* stereochemistry of the trimethylsilylmethyliminium triflate salts has not been commented on, since these salts were generally not isolated in earlier studies. In the present work with cinnamaldehyde imines the salts **2** were readily isolated by stirring cinnamaldehyde *N*-arylimines with trimethylsilylmethyl trifluoromethanesulfonate in diethyl ether and varying mixtures of the *EZ* and *EE* forms were obtained (Scheme 1; Table 1, entries 1–5). The ratios were readily established by the ¹H NMR spectra which showed the styryl α -CH (a doublet of doublets) more upfield when in the shielding region of the *N*-aryl ring in the *EZ* form (Scheme 1). When the salts **2** were desilylated in the presence of the dipolarophiles dimethyl acetylenedicarboxylate (DMAD), *N*-(*p*-substituted phenyl)maleimides, dimethyl fumarate and dimethyl maleate, the respective products **4**, stereoisomer pairs **6,7**, **12,13**, **14,15** (all 1:1 ratio) and **10,11** (1:1 ratio) were obtained in high yields (Table 1). Interestingly the ratio of *EZ* and *EE* isomers in the dipole precursor **2** did not influence the stereoisomer ratio of the cycloadducts, suggesting that the same dipole **3** (*EE* or *EZ*) enters the cycloaddition from either precursor. The 2,5-dihydropyrrole derivatives **4** and the pyrrolidine derivatives **8–11** could be readily oxidised to 2-styrylpyrroles **5** with PbO₂ in CH₂Cl₂. The product series **6, 7, 12–15** are new 6-styryl-substituted 3,7-diazabicyclo[3.3.0]octane-2,4-diones. In all cases the isomeric pairs were readily separated by column chromatography.

All the compounds gave the expected microanalyses and IR and ¹H and ¹³C NMR spectra with the required splitting patterns (Scheme 1 and Experimental section). The stereoisomers were distinguished by the NOE effects of the pyrrolidine 2- and 3-H atoms when these were *cis* and the absence of this NOE effect when these H atoms were *trans* (5-H and 6-H in products **6, 7, 12–15**). In the *exo* cycloadducts **7, 13** and **15** the 6-CH was also significantly more shielded when *cis* to the imido π -electrons, a feature we have also noted¹⁰ for the C-10a bridgehead hydrogen in cycloadducts from the dipole **1**. Varying and unequal *endo:exo* ratios were formed¹⁰ in the cycloadditions of the dipole **1**. The consistent 1:1 ratio of stereoisomers observed herein with the dipole **3** probably reflects the lower steric

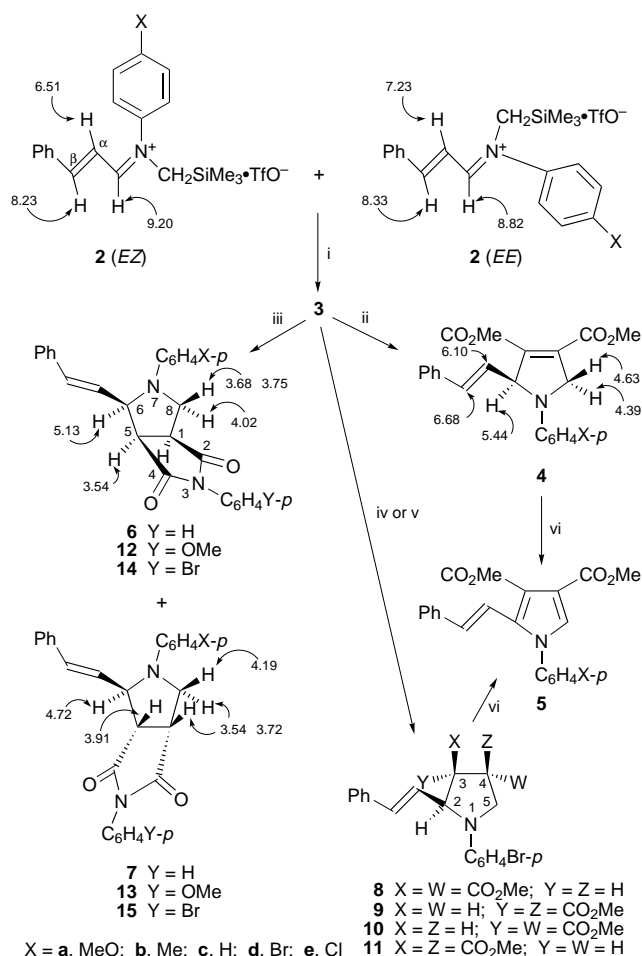


Fig. 1 Reagents: i. CsF; (ii) MeO₂C–C≡C–CO₂Me; iii, *p*-substituted phenylmaleimide; iv, dimethyl fumarate; v, dimethyl maleate; vi, PbO₂ in CH₂Cl₂; some key ¹H NMR shifts (CDCl₃) are shown for X = Br

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Table 1 Substrates and cycloadducts

Entry	Compd.	Mp ^a (T°/C)	Yield (%)	δ_{H}	Compd.	Mp ^a (T°/C)	Yield (%)	δ_{H}
1	2a	158–159 ^b	95(4:1) ^d	—	4a^h	143–145	88	5.38 ^g
2	2b	163–164 ^b	93(1.5:1)	—	4b	104–106	63	5.41 ^g
3	2c	135–137 ^b	92(1:3)	—	4c	99–100	60	5.45 ^g
4	2d	156–158 ^b	90(1.2:1)	—	4d^h	164–166	80	5.44 ^g
5	2e	146–148 ^b	90(1:1)	—	4e	154–155	80	5.40 ^g
6	6d	99–100	40	5.13 ^{e,f}	7d	174–175	40	4.72 ^e
7	6c	198–200 ^c	40	5.19 ^{e,f}	7c	173–174 ^c	40	4.75 ^e
8	12d	120–122	30	5.10 ^{e,f}	13d	190–192	30	4.71 ^e
9	14d	178–180	46	5.09 ^{e,f}	15d	152–154	46	4.70 ^e
10	8	84–85	35	4.58 ^{f,g}	9^h	88–89	35	4.66 ^{f,g}
11	10	124–126	20	4.59 ^g	11	130–132	20	4.62 ^{f,g}

^aFrom EtOH unless stated otherwise. ^bWashed with Et₂O. ^cFrom MeOH. ^dParentheses contain *EZ:EE* ratio. ^e δ_{H} for H-6. ^fNOE enhancement from adjacent *cis* H. ^g δ_{H} for H-2. ^hOxidation gave compounds **5a** and **5d**.

strain and greater flexibility in the cycloaddition transition state.

Experimental

Mps were measured on an Electrothermal apparatus. IR spectra were measured with a Perkin Elmer 983G spectrophotometer. NMR spectra were measured on JEOL GX FT270 and GXFT400 spectrometers using CDCl₃ as solvent. Cinnamaldehyde *N*-arylimines separated on cooling after a dry EtOH solution of the aldehyde and the *p*-substituted aniline (1:1 mol) was heated under reflux for 1 h. The salts **2** were obtained in 90–95% yields when equimolar quantities of the imine and trimethylsilylmethyl triflate were stirred in dry Et₂O at ambient temperature for 12 h.

Cycloadditions (Typical Examples).—3,7-Diphenyl-6-exo-[(*E*-styryl)-3,7-diazabicyclo[3.3.0]octane-2,4-dione **7c** and the endo isomer **6c** (Table 1, entry 7). A solution of the triflate salt **2c** (0.50 g, 1.38 mmol) in dry CH₂Cl₂ (20 cm³) was treated with *N*-phenylmaleimide (0.48 g, 2.76 mmol) followed by an excess of CsF. The mixture was stirred under anhydrous conditions at ambient temperatures for 24 h and then filtered. The filtrate (together with the CH₂Cl₂ filter-cake washings) was evaporated under reduced pressure to 4 cm³, placed on a column of silica gel (230–400 mesh) and eluted slowly with CH₂Cl₂. First eluted was the endo isomer **6c** (40%) mp 198–200 °C (from MeOH) (Found: C, 79.0; H, 5.5; N, 7.0. C₂₇H₂₂NO₂ requires C, 79.2; H, 5.6; N, 7.1%) $\nu_{\text{max}}/\text{cm}^{-1}$ (mull) 1709 br (amido C=O); δ_{H} (CDCl₃) 3.54 (d, *J* 7.3 Hz, H-5), 3.64–3.79 (m, 2 H, H-1, H-8_{endo}), 4.08 (d, *J* 9.5 Hz, H-8_{exo}), 5.19 (d, *J* 6.6 Hz, 1 H, H-6), 6.08–6.16 (dd, 1 H, styryl, H α), 6.62 (d, *J* 15.4 Hz, 1 H, styryl, H β), 6.74–6.8 (m, 3 H, Ar), 7.21–7.49 (m, 12 H, Ar); δ_{C} (CDCl₃) (off-res) 43.6 (d, C-1), 49.6 (t, C-8), 51.5 (d, C-5), 63.3 (d, C-6), 145.1, 115.1, 128.5, 118.7 (N-7-Ph, C-1', C-2', C-3', C-4' resp.), 135.9, 126.6, 129.3, 125.7 (styryl phenyl, C-1', C-2', C-3', C-4' resp.), 131.5, 126.5, 128.7, 128.2 (N-3-Ph, C-1', C-3', C-4' resp.), 128.9 (d, styryl C- α), 133.2 (d, styryl C- β), 176.8, 177.9 (s, C=O).

Next eluted was the exo isomer **7c** (40%) mp 173–174 °C (from MeOH) (Found: C, 79.1; H, 5.4; N, 6.9. C₂₇H₂₂NO₂ requires C, 79.2; H, 5.6; N, 7.1%) $\nu_{\text{max}}/\text{cm}^{-1}$ (mull) 1712 br (amido C=O); δ_{H} (CDCl₃) 3.68–3.76 (m, 2 H, H-8_{endo}, H-1), 3.85–3.92 (m, 1 H, H-5), 4.23–4.26 (m, 1 H, H-8_{exo}), 4.75 (m, 1 H, H-6), 6.17–6.26 (dd, *J* 16.1, 5.9 Hz, 1 H, styryl, H- α), 6.56 (d, 1 H, styryl, H β), 6.78–6.86 (m, 3 H, Ar), 7.03–7.07 (2 H, m, Ar), 7.22–7.30 (10 H, m, Ar); δ_{C} (CDCl₃) (off-res). 44.9 (d, C-1), 49.9 (d, C-5) 50.9 (t, C-8), 62.3 (d, C-6), 146.4, 115.3, 128.6, 119.1 (N-7-Ph, C-1', C-2', C-3', C-4' resp.), 135.9, 126.6, 129.2, 126.8 (styryl phenyl C-1', C-2', C-3', C-4' resp.), 131.6, 124.9, 129.2, 128.1 (N-3-Ph, -1', C-2', C-3', C-4' resp.), 128.7 (d, styryl C- α) 133.1 (d, styryl, C- β), 174.8, 177.3 (s, C=O).

1-(*p*-Bromophenyl)-3,4-bis(methoxycarbonyl)-2-[(*E*-styryl)-2,5-dihydropyrrole **4d** and pyrrole **5d** (Table 1, entry 4).—A solution of the triflate salt **2d** (0.50 g, 1.13 mmol) in dry CH₂Cl₂ (20 cm³) was treated with dimethyl acetylenedicarboxylate (0.28 ml, 2.6 mmol) followed by an excess of CsF. The mixture was stirred under anhydrous conditions at ambient temperatures for 24 h and then filtered. The filtrate (together with the CH₂Cl₂ filter-cake washings) was evaporated under reduced pressure to 4 cm³, placed on a flash column of silica gel (230–400 mesh ASTM) and eluted with gradient mixtures of petroleum spirit (bp 40–60 °C)–CH₂Cl₂ (1:0 to 1:1) to give the dihydropyrrole **4d** (80%), mp 164–166 °C (from EtOH)

(Found C, 59.5; H, 4.7; N, 3.1. C₂₂H₂₀BrNO₄ requires C, 59.7; H, 4.5; N, 3.1.1%; $\nu_{\text{max}}/\text{cm}^{-1}$ (mull) 1725, 1730 (ester C=O); δ_{H} (CDCl₃) 3.81 (s, 3 H, CO₂Me), 3.85 (s, 3 H, CO₂Me), 4.39 (dd, 1 H, H-5 *trans* relative to styryl), 4.63 (dd, 1 H, H-5 *cis*), 5.44 (m, 1 H, H-2), 6.10 (dd, 1 H, styryl, H- α), 6.49–6.53 (m, 2 H, N-ArBr), 6.68 (d, *J* 16.1 Hz, 1 H, styryl, H- β), 7.27–7.32 (m, 7 H, Ar); δ_{C} (CDCl₃) (off-res) 52.6 (q, OMe), 55.9 (t, C-5), 69.3 (d, C-2), 144.6, 113.9, 132.0, 109.5 (N-1-*p*-BrC₆H₄, C-1', C-2', C-3', C-4' resp.), 139.9, 126.8, 128.2, 126.2 (styryl phenyl, C-1', C-2', C-3', C-4' resp.), 132.4 (s, C-4), 135.9 (s, C-3), 128.1 (d, styryl C- α), 135.9 (d, styryl, C- β), 162.9, 163.5 (s, C=O).

Oxidation of **4d** (0.45 mmol) with PbO₂ (4.5 mmol) in CH₂Cl₂ (20 cm³) for 48 h at ambient temperature gave the pyrrole **5d** (50%), mp 94–95 °C (from EtOH) (isolated by elution from a flash column of silica gel 230–400 mesh ASTM with petroleum spirit (bp 40–60 °C)–CH₂Cl₂ (20:1 v/v) (Found: C, 59.7; H, 4.0; N, 3.2. C₂₂H₁₈BrNO₄ requires C, 60.0; H, 4.1; N, 3.2%) $\nu_{\text{max}}/\text{cm}^{-1}$ (mull), 1724, 1710 (ester C=O); δ_{H} (CDCl₃) 3.85 (s, 3 H, OMe), 3.93 (s, 3 H, OMe), 6.79 (dd, 2 H, N-1-*p*-BrC₆H₄, H-2'), 7.23–7.35 (m, 9 H, aromatic protons, styryl, H α , H β), 7.64 (m, 1 H, H-5); δ_{C} (CDCl₃) (off-res.), 51.7, 52.4 (q, OMe), 137.5, 115.3, 132.9, 116.3 (N-1-*p*-BrC₆H₄, C-1', C-2', C-3', C-4' resp.), 136.5, 127.8, 128.7, 126.5 (styryl phenyl, C-1', C-2', C-3', C-4' resp.), 122.6 (s, C-4), 127.9 (s, C-3), 128.0 (d, styryl, C- α), 128.5 (d, styryl, C- β), 132.7 (C-2), 133.5 (d, -5), 163.8, 166.3 (s, C=O).

The 1-(*p*-methoxyphenyl) pyrrole **5a**, mp 104–106 °C (from EtOH), was similarly obtained.

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